ANALYSIS OF NITROGEN REMOVAL PERFORMANCE OF CONSTRUCTED RAPID INFILTRATION SYSTEM (CRIS)

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(Received 3rd Aug 2016; accepted 3rd Oct 2016)

Abstract. The Constructed Rapid Infiltration System (CRIS) is widely applied for wastewater treatment. In China, however, fundamental research on functionality of CRIS is lacking. We used a CRIS simulation column to treat domestic sewage under experimental conditions, allowing determination of nitrogen pollutants removal performance. The obtained results showed that CRIS can effectively remove ammonia nitrogen with the average removal rate of 82.2% but results in a relatively low removal of total nitrogen (TN) with the average removal rate of 31.9%. The reason is that the system is frequently subjected to alternation of drying and wetting, which allows effective aeration. This leads to aerobic conditions, which is incompatible with the anaerobic requirements of denitrifying bacteria. In addition, these bacteria likely lack a suitable carbon source, as organic matter is effectively removed in the upper (aerobic) layer of CRIS. Lastly, the filter material used is generally negatively charged, thus repels nitrate, which prevents its retention and results in its discharge with the outflow. As a net result, the TN removal rate is low. **Keywords:** *ammonia nitrogen; TN; nitrate nitrogen; organic matter; removal performance and mechanism*

Introduction

The Constructed Rapid Infiltration System (CRIS) was developed for ecologicallyfriendly treatment of sewage and contaminated surface water, based on a rapid and simple infiltration system, by professor Zhong of China University of Geosciences (Xu et al., 2015; Xu et al., 2013). Since its development, it has been widely applied to wastewater treatment in domestic settings. The system is composed of a grille, a preliminary sedimentation basin, a high-permeability basin containing filter material and an outflow system. The high-permeability basin contains a fixed amount of artificial filter material, allowing to operate under alternating dry-wet conditions with a capacity of 1.0-1.5 m^3/m^2d or m/d; the system is typically subjected to frequent flooding with intermittent dry periods (Xie et al., 2010). Wastewater is purified through adsorption, interception and decomposition of aquatic pollutants by microorganisms that are naturally present in the filter material of CRIS. The unique structure and dry-wet alternate inflow mode of CRIS result in a diverse microbial growth on the surface of the percolation medium, so that the medium usually supports both aerobic, anaerobic, and facultative anaerobic bacteria, all of which contribute to effective treatment of wastewater (Xu et al., 2011b; Liu, 2006). CRIS is preferably applied to treat domestic sewage from small towns as well as contaminated surface water. Removal of COD is typically about 85-90%, ammonia nitrogen can be removed to over 90%, and SS and LAS removal is achieved above 95% (Jiang et al., 2011; Ma et al., 2008). With its convenient operation management and low investment and maintenance costs, CRIS is

successfully promoted in many districts of China, with obvious benefits for the community, local economy and wildlife (Xu et al., 2011a; Liu, 2006). Nevertheless, at present most results on effectiveness are empiric, while basic and theoretical research is lacking. The research presented here was conducted to determine the efficiency and robustness of nitrogen removal by means of experiments performed using CRIS simulation columns processing domestic sewage. A theoretical interpretation of the biological processes at play is provided.

Experimental materials and method

Experimental device

A CRIS simulation column was constructed in the lab for research purposes. The main body of the reactor was composed of a hard PVC pipe of 200 cm high and an internal diameter of 20 cm, the filter material consisted of 85% natural sand, 5% ferrous powder, 5% marble sand and 5% zeolite sand, and the height of the filter layer was 150 cm. There was a sampling port every 0.25 m from top to bottom of the filter layer. Water was allowed to flow from top to bottom. A schematic of the experimental installation is shown in *Fig.1*.



Figure 1. Experimental installation

Monitoring Variables

In order to resemble natural conditions, domestic wastewater was used as the experimental water sample, with quality indicators as shown in *Table 1*.

Table 1. Raw water of	quality indicators
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<u>WaterTemperature</u>	рН	COD	SS	TN	TP	NH4 ⁺ -N
(⁰ C)		(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
17-25	7.00-8.10	130-160	30-60	30-48	0.90-1.80	20-38

CRI was started up with raw wastewater. When the COD removal rate became stable, indicative of a successful and stable biofilm being formed, the CRIS analog column was fed with a hydraulic load of $1.00 \text{ m}^3/\text{m}^2$ per day. CRIS analog column was fed wastewater once every six hours and each feed lasted 20 minutes. To study nitrogen

removal performance, the residual concentrations of nitrogen pollutants at various steps of the process were measured from water tapped from the sampling points and from the effluent once every two days. The experiment was performed from April 8, 2015 to June 30, 2015. The analytical methods used in the study are described in Methods of Water and Wastewater Monitoring and Analysis (edition 4) (Wang, 2002). The used tests determined COD_{Cr} levels by the Potassium dichromate method, NH₃-N by Nessler reagent colorimetry, NO₃-N by ultraviolet (UV) spectroscopy and total nitrogen (TN) by alkaline potassium sulfate digestion followed by UV spectrophotometry. Since nitrogen in the form of nitrite was relatively low in the inflow water and was previously shown to be difficult to accumulate by CRIS (Liu, 2006), this variable was not determined.

Results

Nitrogen Removal Principal of CRIS

The CRIS took 14 cycles before outflow was stable. Once stability was achieved, the concentrations of nitrogen pollutants in the inflow and effluent of the experimental CRIS were determined as shown in *Table 2*. The average removal of nitrogen present in the form of ammonia was 82.2%, while TN was removed with 31.9% efficiency on average. In contrast, concentrations of nitrogen in the form of nitrate increased from 2.4mg/L in the inflow to 12.5mg/L in the effluent (*Table 2*). These results were obtained with inflow water in which the ammonia concentration varied considerably, between 20.9mg/L and 35.8mg/L. This variation was approximately halved in the effluent, suggesting that the system is highly resistant to variation in ammonia nitrogen input. The amplitude of variation in TN removal was relatively large. As a result, the concentrations of nitrate nitrogen and TN in the effluent varied considerably (*Table 2*).

	Ammonia Nitrogen			Т	'otal Nitrog	Nitrate Nitrogen		
No.	inflow (mg/L)	effluent (mg/L)	removal (%)	inflow (mg/L)	effluent (mg/L)	removal (%)	inflow (mg/L)	effluent (mg/L)
14	31.2± 1.1	$5.5\pm$ 0.2	82.5	43.6± 1.9	29.9± 1.1	31.4	$2.3\pm$ 0.09	$\begin{array}{c} 10.4 \pm \\ 0.3 \end{array}$
15	27.8± 1.2	$4.9\pm$ 0.2	82.4	$\begin{array}{r} 40.5 \pm \\ 2.0 \end{array}$	$27.5\pm$ 1.2	32.2	$1.9\pm$ 0.06	$12.1\pm$ 0.5
16	$\begin{array}{c} 32.5 \pm \\ 1.6 \end{array}$	$6.2\pm$ 0.1	80.9	$\begin{array}{c} 42.9 \pm \\ 1.8 \end{array}$	$\begin{array}{c} 29.7 \pm \\ 1.3 \end{array}$	30.9	$2.4\pm$ 0.11	$11.3\pm$ 0.3
17	$31.8\pm$ 1.5	$4.1\pm$ 0.2	87.1	$\begin{array}{c} 40.5 \pm \\ 1.9 \end{array}$	$29.9\pm$ 1.4	33.7	$2.3\pm$ 0.11	$12.6\pm$ 0.4
18	33.6± 1.6	$6.3\pm$ 0.3	81.3	$\begin{array}{r} 43.8 \pm \\ 2.1 \end{array}$	$30.9\pm$ 1.4	29.4	$2.4\pm$ 0.11	$\begin{array}{c} 10.5 \pm \\ 0.2 \end{array}$
19	$\begin{array}{c} 28.1 \pm \\ 1.3 \end{array}$	$5.5\pm$ 0.2	80.6	38.6± 1.6	$28.7\pm$ 1.1	33.5	$1.8\pm$ 0.08	$13.4\pm$ 0.3
20	$23.9\pm$ 1.0	$4.5\pm$ 0.1	81.1	$34.2\pm$ 1.2	$25.9\pm$ 1.2	30.2	$2.3\pm$ 0.03	12.3 ± 0.4
21	$21.7\pm$ 0.8	$4.2\pm$ 0.2	80.5	$30.9\pm$ 1.5	$23.3\pm$ 1.0	27.8	$2.6\pm$ 0.12	$12.5\pm$ 0.1
22	$34.8\pm$	$6.2\pm$	82.3	$42.5\pm$	$27.0\pm$	36.4	$1.6\pm$	$13.5\pm$

Table 2. Inflow and effluent measurements

APPLIED ECOLOGY AND ENVIRONMENTAL RESEARCH 15(1): 199-206. http://www.aloki.hu • ISSN 1589 1623 (Print) • ISSN 1785 0037 (Online) DOI: http://dx.doi.org/10.15666/aeer/1501_199206 © 2017, ALÖKI Kft., Budapest, Hungary

	1.2	0.1		2.1	1.2		0.05	0.1
23	$27.6\pm$	$4.6\pm$	83.2	$39.4\pm$	$27.1\pm$	31.2	$2.9\pm$	$12.4\pm$
25	1.2	0.2	05.2	1.8	1.3	51.2	0.13	0.5
24	$20.9\pm$	$3.8\pm$	817	$32.6\pm$	$23.3\pm$	28.5	$2.5\pm$	$11.6\pm$
27	0.6	0.1	01.7	1.4	0.9	20.5	0.10	0.1
25	$26.9\pm$	$4.7\pm$	82 /	$37.8\pm$	$26.3\pm$	30.5	$3.1\pm$	$14.4\pm$
23	1.1	0.2	02.4	1.6	1.2	50.5	0.15	0.2
26	$27.7\pm$	$5.2\pm$	81.1	$37.6\pm$	$27.7\pm$	34.4	$2.4\pm$	$13.7\pm$
20	1.3	0.2	01.1	1.5	1.3	54.4	0.10	0.2
27	$35.8\pm$	$6.7\pm$	81.2	$45.5\pm$	$30.6\pm$	34.9	$2.6\pm$	$13.3\pm$
21	1.5	0.3	01.2	2.1	1.4	54.9	0.11	0.4
28	$33.1\pm$	$5.8\pm$	82 /	$41.5\pm$	$28.6\pm$	31.1	$1.7\pm$	$12.8\pm$
20	1.6	0.2	02.4	2.0	0.8	51.1	0.04	0.1
20	$26.9\pm$	$4.8\pm$	82.1	$37.8\pm$	$26.2\pm$	30.6	$2.5\pm$	$12.2\pm$
29	0.9	0.1	02.1	1.8	1.1	50.0	0.10	0.1
30	$29.8\pm$	$5.9\pm$	80.1	$41.5\pm$	$28.4\pm$	317	$3.3\pm$	$13.8\pm$
50	1.3	0.2	00.1	1.9	1.1	51.7	0.15	0.3
31	$31.7\pm$	$5.6\pm$	823	$43.8\pm$	$28.4\pm$	35.2	$2.8\pm$	$11.9\pm$
51	1.5	0.2	02.5	1.8	1.2	55.2	0.13	0.1
32	$32.1\pm$	$4.6\pm$	857	$44.3\pm$	$29.5\pm$	33 /	$2.3\pm$	$13.0\pm$
52	1.2	0.1	05.7	2.1	1.3	55.4	0.09	0.5
Mean value	29.4	5.24	82.2	40.0	27.8	31.9	2.4	12.5

Liu (2006) concluded that the top 100cm of CRIS typically represent an aerobic zone, while the 30 cm below this would be an anoxic zone. As can be seen from *Table 3* and Figure 2, presenting findings from the individual sample points, the removal of ammonia nitrogen in the top section of CRIS (0-25 cm) was 9%, increasing to 78% accumulatively when the water had penetrated 100cm, after which the removal of ammonia nitrogen levelled off to 84%. The highest reduction increments were seen between 25cm, 50cm and 75cm, thus, most ammonia nitrogen is removed in the filtering between 25-75cm, not in the top 0-25cm. Possibly, the concentration of organic matter in the inflow promoted growth of heterotrophic bacteria, and this would inhibit the growth of nitrobacteria in the top layer of the filter (Hu et al., 2010). Related studies have shown that within a reactor there is competition for space to form biofilms as well as competition for dissolved oxygen among microorganisms, especially for the aerobic population (Hu et al., 2010). Nitrobacteria are chemo-autotrophs that, due to their lower growth rates, are generally outcompeted by heterotrophic bacteria. Moreover, dissolved oxygen will be abundant in the top layer of the column, where it is utilized by the surface biofilm consisting of heterotrophic bacteria residing there (Liu, 2006). By the time dissolved oxygen is diffused to where it can be utilized by nitrobacteria, its concentration would have decreased considerably, further limitating the reproduction of nitrobacteria (Xu et al., 2011a). Finally, in the early stage of inflow, most organic nitrogen was decomposed to ammonia nitrogen, which increased the concentration of ammonia nitrogen, further contributing to the low ammonia nitrogen removal in the top section of 0-25cm of CRIS (Xu et al., 2011b).

Table 4 and *Figure 2* lists the total nitrogen concentrations at individual sample points. As can be seen, TN removal in filtering layer 0-75cm was 13.4% on average, decreasing to 11.4%(24.8%-13.4%) in the section 75-100cm and to 6.5%(31.3%-24.8%) for the last 100-150cm. Nitrate nitrogen increased by 1.95mg/L in the section 0-25cm,

and by 3.99mg/L in section 25-50cm, where it increased at the highest rate (*Table 5* and *Figure 2*). After this, the increment dropped to 3.11mg/L in layer 50-100cm, followed by marginal removal of 0.41mg/L in layer 100-150cm. Since ammonia nitrogen was the main form of nitrogen in the inflow, the nitrate nitrogen concentration is comparatively low, so the ammonia nitrogen was nearly completely nitrificated in layer 0-100cm, which increased the nitrate nitrogen concentration in this section (Ma et al., 2009). Removal of ammonia nitrogen was high in the top 100cm and low in the section 100-150cm, while nitrification occurring earlier had consumed most of the dissolved oxygen, resulting in inhibition of nitrification in layer 100-150cm. Instead, denitrification was promoted here by anaerobic bacteria. Taking together, nitrate nitrogen increased steadily in layer 100-150cm (Xu et al., 2011c).

	Depth of CRIS Filtration Pool(cm)								
No.	0	25	50	75	100	125	150		
43	$30.5\pm$	$27.4\pm$	$19.3\pm$	9.6 ± 0.4	6.1 ± 0.3	5.0 ± 0.2	4.3 ± 0.2		
	1.2	1.1	0.9						
44	$31.0\pm$	$27.9\pm$	$19.7\pm$	$10.2\pm$	6.3 ± 0.3	5.5 ± 0.1	4.9 ± 0.2		
	1.5	1.2	0.7	0.5					
45	$31.7\pm$	$28.2\pm$	$20.0\pm$	$10.7\pm$	6.7 ± 0.2	5.8 ± 0.2	5.2 ± 0.1		
	1.4	1.1	0.7	0.5					
46	$29.3\pm$	$27.4\pm$	$18.4\pm$	9.2 ± 0.3	6.8 ± 0.1	5.4 ± 0.1	4.6 ± 0.2		
	0.9	1.2	0.6						
47	$32.6\pm$	$29.3\pm$	$19.8\pm$	$11.3\pm$	7.3 ± 0.3	5.8 ± 0.2	5.2 ± 0.1		
	1.2	0.8	0.5	0.4					
Mean	31.0	28.0	19.4	10.2	6.6	5.5	4.8		
Concentration									
Mean	0	9.6	37.3	67.1	78.6	82.3	84.4		
Removal(%)									

 Table 3. Ammonia nitrate concentrations at individual sample points

Table 4. Total nitrogen concentrations at individual sample points

	Depth of CRIS Filtration Pool(cm)								
No.	0	25	50	75	100	125	150		
43	$43.5\pm$	$41.9\pm$	$39.7\pm$	37.6±	$32.3\pm$	$30.5\pm$	$29.3\pm$		
	2.1	1.1	1.9	1.8	1.6	1.5	1.4		
44	$42.0\pm$	$40.8\pm$	$38.8\pm$	$36.8\pm$	$32.5\pm$	$30.2\pm$	$29.6\pm$		
	2.0	0.7	1.8	1.8	1.6	1.2	1.4		
45	$40.1\pm$	$38.5\pm$	$37.5\pm$	$35.7\pm$	$31.0\pm$	$29.8\pm$	$29.4\pm$		
	1.1	1.9	1.8	1.6	1.5	1.4	1.3		
46	$45.2\pm$	$43.3\pm$	$42.0\pm$	$38.9\pm$	$33.4\pm$	$31.7\pm$	$28.5\pm$		
	2.0	1.7	2.0	1.7	1.6	1.2	1.2		
47	$42.3\pm$	$40.7\pm$	$38.5\pm$	$35.5\pm$	$31.1\pm$	$29.9\pm$	$29.6\pm$		
	1.8	2.0	1.6	1.6	1.5	1.4	1.3		
Mean	42.6	41.0	39.3	36.9	32.0	30.4	29.2		
Concentration									
Mean	0	3.7	7.7	13.4	24.8	28.6	31.3		
Removal(%)									

	Depth of CRIS Filtration Pool(cm)									
No.	0	25	50	75	100	125	150			
43	$2.50\pm$	$5.74\pm$	$8.58\pm$	$10.07\pm$	$11.10\pm$	$11.68\pm$	$11.72\pm$			
	0.12	0.26	0.42	0.39	0.55	0.43	0.57			
44	$2.20\pm$	$5.60\pm$	$8.82\pm$	$9.89\pm$	$11.95\pm$	$12.06\pm$	$12.12\pm$			
	0.09	0.25	0.41	0.47	0.57	0.60	0.60			
45	$2.71\pm$	$5.93\pm$	$7.95\pm$	$9.62\pm$	$11.88\pm$	$12.09\pm$	$12.11\pm$			
	0.13	0.29	0.39	0.44	0.56	0.48	0.59			
46	$3.15\pm$	$6.75\pm$	$8.89\pm$	$10.19\pm$	$12.38\pm$	$12.54\pm$	$12.97\pm$			
	0.28	0.33	0.27	0.48	0.55	0.52	0.61			
47	$2.40\pm$	$6.12\pm$	$8.45\pm$	$9.95\pm$	$11.24\pm$	$11.66\pm$	$11.71\pm$			
	0.10	0.29	0.37	0.42	0.45	0.54	0.52			
Mean	2.59	6.03	8.54	9.94	11.71	12.01	12.13			
Concentration										

Table 5. Nitrate nitrogen concentrations at individual sample points



Figure 2. Pollutant concentration and the removal rates at individual sample points

Denitrification Mechanism Analysis of CRIS

Ammonia is mainly oxidized to nitrate by nitrosifying and nitrifying bacteria in CRIS (Zhao, 2010). It is first oxidized to nitrite by nitrosifying bacteria, which is then oxidized to nitrate nitrogen by nitrofying bacteria. The relative amounts and activities of nitrosifying and nitrifying bacteria affect the nitrofying process directly (Xu, 2011). Both populations depend on environmental factors for their survival such as pH, temperature, absence of toxic and other harmful substances such as heavy metals and dissolved oxygen (Xu, 2011). The experiments described here modelled treatment of domestic sewage, which contains very low concentrations of heavy metals. During the experiment the room temperature varied between 20-30°C, which is close to the optimal temperature for nitrification. The filtrate contained 10% marble sand that maintained the pH between 7.15 and 8.03, close to the optimal pH range for nitrification. We conclude that dissolved oxygen was the main factor affecting the nitrification in CRIS (Xu, 2011).

The reason why TN removal was low in CRIS can be explained as follows. Ammonia nitrogen ions are positively charged and are absorbed by negatively-charged filter particles and microorganisms. The ions are transformed into nitrate by nitrifying bacteria under aerobic conditions, which can further be metabolized to gaseous nitrogen (N_2) under anaerobic conditions, resulting in nitrogen removal from the system (Xu, 2011). *Table 5* shows that in filter chamber, the oncentration of nitrate nitrogen was above 12mg/L in outflow, compared to approximately 2mg/L in the inflow, which suggested that nitrification dominated denitrification. Denitrifying bacteria use nitrate nitrogen as electron acceptor and organic matter as donor to conduct anaerobic respiration, with the net result that nitrate nitrogen is reduced to gaseous nitrogen. Nitrate nitrogen denitrification can only be achieved under strict anaerobic conditions, provided there is enough organic carbon present (Xu, 2011; Yao, 2006).

CRIS is used under conditions where it is frequently flooded with intermittent dry periods, which results in efficient aeration. Since nitrate nitrogen concentration increased in CRIS (*Table 5* and *Figure 2*), it can be concluded that aerobic conditions is applied, since denitrifying bacteria are strictly anaerobic. Nitrate nitrogen is not easily converted to nitrogen gas via denitrification. In deeper layers (100-150cm) of the filter chamber, the concentration of dissolved oxygen was low, but organic matter had most likely been decomposed and removed in the layers above, since CRIS is effective for organic matter removal (*Table 6* and *Figure 2*). As such, denitrification was most probably inhibited by lack of a suitable carbon source, and nitrate nitrogen couldn't be transformed into nitrogen gas to be removed. Nitrate nitrogen is negatively charged and dissolves well in water; therefore, it cannot be absorbed or retained by filter material or microorganisms. Instead, it maintains dissolved in the water and is removed with the outflow, resulting in a relatively high concentration of TN in the outflow of CRIS.

	Depth of CRIS Filtration Pool(cm)							
	0	25	50	75	100	125	150	
Mean Concentration	160.10	90.87	61.67	36.23	30.82	23.33	19.69	
Mean Removal(%)	0	43	61	77	81	85	88	

Table 6. COD concentrations at individual sample points

Conclusions

There are three reasons why removal of ammonia nitrogen but not of total nitrogen is effective in CRIS:

(1) The lack of anaerobic conditions: frequent dry periods intermitting flooding result in effective aeration, and that leads to operation under aerobic conditions, which cannot support denitrifying bacteria.

(2) The lack of organic matter: organic matter is almost completely removed in the upper layer of CRIS, further contributing to low denitrification efficiency as the denitrifying bacteria lack a carbon source.

(3) Filter material has the same charge as nitrate: both are negatively charged and nitrate nitrogen is highly soluble in water, which makes it difficult to retain nitrate in the filter material; instead it is discharged with the outflow resulting in a net low TN removal rate.

Acknowledgements. The research was funded by the Natural Science Foundation of China (No. 41502333),the State Key Laboratory of Geohazard Prevention and Geoenvironment Protection Foundation (No.SKLGP2015Z012, SKLGP2014Z001), the specialized research fund for the doctoral program of colleges and universities (No.20135122120020), the scientific research plan of education department of Sichuan Province (No.14ZB0073).

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